

CHEMISTRY OF DIHYDRO-1,4-DIOXIN III <sup>1</sup>  
A NEW METHOD FOR THE PREPARATION OF  $\alpha, \alpha'$ -DIHYDROXY  
KETONES FROM KETONES AND ALDEHYDES

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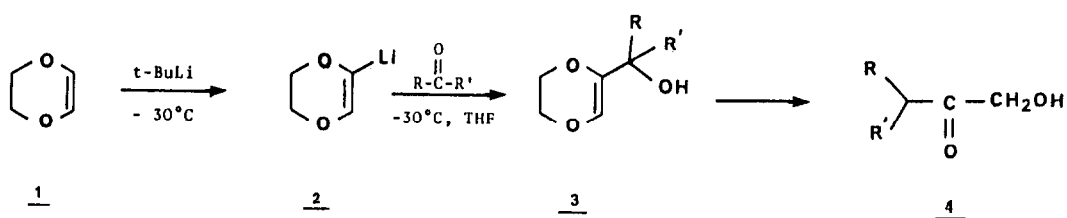
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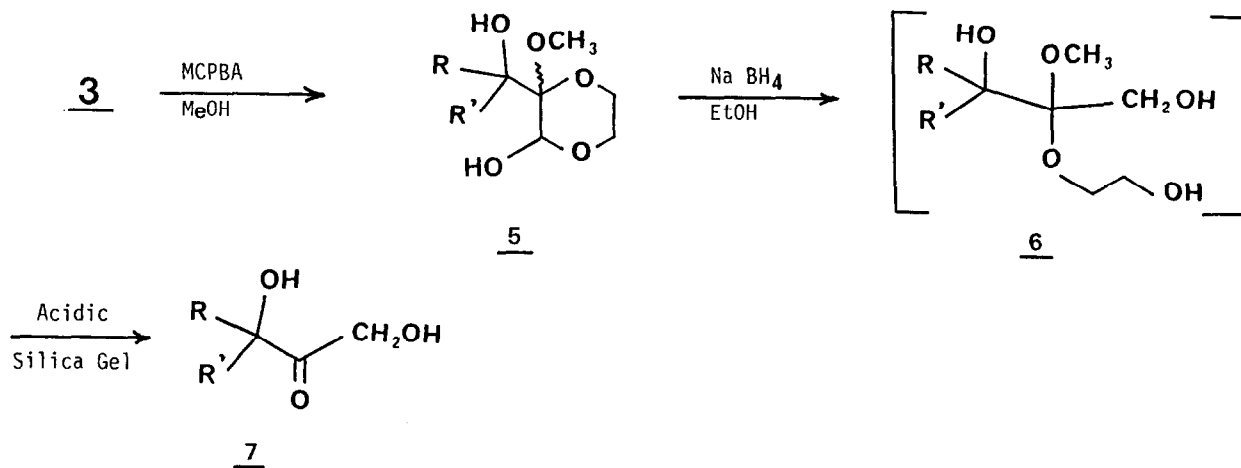
*Summary* : Title compounds, 7, were prepared from ketones and aldehydes via the intermediate 3 by methanolic peracid epoxidation followed by  $\text{NaBH}_4$  reduction and acidic hydrolysis. Application of this method to the preparation of the unnatural corticoid side chain was reported.

Dihydro-1,4-dioxin-2-yl lithium 2, prepared by lithiation of dihydro-1,4-dioxin 1 with t-butyl lithium, reacts with ketones and aldehydes to give alcohols 3. These adducts, which are moderately stable, undergo various transformations and are interesting synthetic intermediates. In a previous communication, we have described a method for the preparation of  $\alpha$ -hydroxy ketones 4 from ketones and aldehydes via these intermediates (scheme I)<sup>1</sup>. We now report a new procedure, starting from the same alcohols 3, for the preparation of  $\alpha, \alpha'$ - dihydroxy ketones 7.

S c h e m e I



## Scheme II



Table

Entry	R	R'	<u>3</u>	Yield %	<u>5</u>	Yield %	$\alpha, \alpha'$ -dihydroxy ketone <u>7</u> <sup>6</sup>	Yield % <sup>**</sup>
a		H	<u>3a</u>	52	<u>5a</u>	55		65
b			<u>3b</u>	56	<u>5b</u>	55		45
c			<u>3c</u>	54	<u>5c</u>	63		60
d		H	<u>3d</u>	60	<u>5d</u>	75		70
e			<u>3e</u>	64	<u>5e</u>	82		90

<sup>\*\*</sup> The yields (not optimized) are given for the pure isolated products .

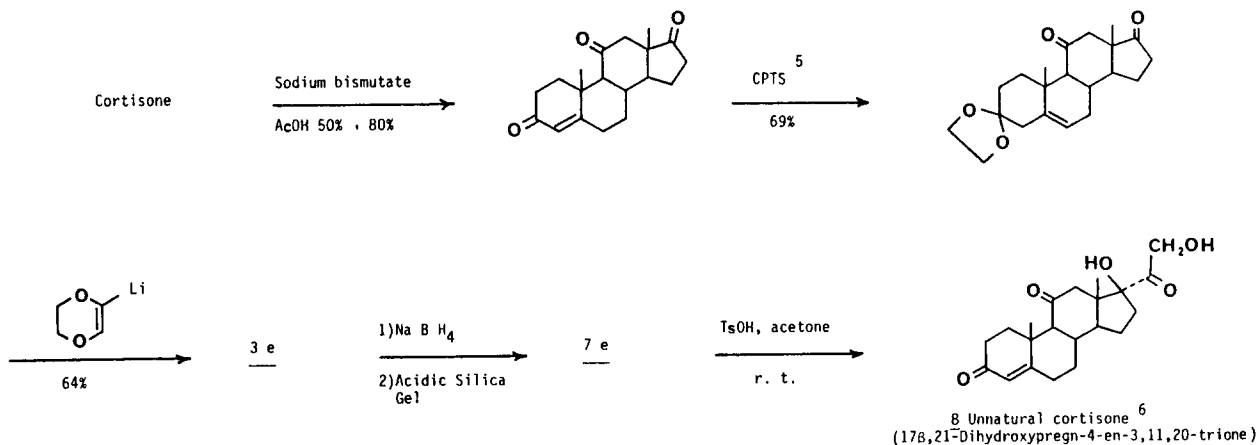
As in 2,3-dihydropyran 2 and enol ethers 3, allylic alcohols 3 smoothly undergo epoxidation with *m*-chloroperbenzoic acid in methanol at 0°C leading to the  $\alpha$ -ketoaldehyde equivalent 5 (Scheme II). <sup>1</sup>H and <sup>13</sup>C NMR spectra showed that 5, although homogenous in TLC, consisted of a mixture of two diastereoisomers. It is noteworthy that the epoxidation proceeded in a highly regioselective manner: reaction always occurred on the dioxene group double bond when another double bond is present in the molecule (Table, entries d and e).

Sodium borohydride reduction of 5 and subsequent hydrolysis with wet silica gel <sup>4</sup> afforded  $\alpha, \alpha'$ -dihydroxy ketones 7 <sup>6</sup> in fair yields (see table).

A typical experiment is as follows: adduct 3d (0,6g,  $\approx$  3mmol) in methanol (15ml) was epoxidized by known procedure <sup>2</sup> except that methanol was evaporated under reduced pressure before extraction with CH<sub>2</sub>Cl<sub>2</sub>. The crude product purified by flash chromatography (1:1 ethyl acetate-petroleum ether) affording 5d as a diastereomeric mixture (0,51g, 75% yield). Understirring 5d (456 mg, 2mmol) in ethanol (15 ml) was reduced by solid NaBH<sub>4</sub> (75mg) for 1h. Evaporation of ethanol under reduced pressure and extraction with ethyl acetate, gave crude 6d which was hydrolyzed with wet silica gel (2g) in CH<sub>2</sub>Cl<sub>2</sub> (6ml) and 10% aqueous oxalic acid (7 drops) <sup>4</sup>. Flash chromatography of the crude product afforded pure  $\alpha, \alpha'$ -dihydroxy ketone 7d (240 mg, 70% yield).

The utilization of the present reaction sequence, a conversion of carbonyl compounds to  $\alpha, \alpha'$ -dihydroxy ketones, is illustrated by the stereoselective synthesis of 17-epicortisone 8 with unnatural stereochemistry (Scheme III).

Scheme III



## References and Notes

- 1 . For part II, see M. FETIZON, I. HANNA and J. RENS, *Tetrahedron Lett.* 1985, in the press .
- 2 . A.A. FRIMER, *Synthesis* 1977, 578 .
- 3 . F. HUET, A. LECHEVALLIER, J.M. CONIA, *Synthet. Commun.* 1980, 10, 83.  
G.M. RUBOTTOM, R. MARRERO, *ibid.* 1981, 11, 505.
- 4 . F. HUET, A. LECHEVALLIER, M. PELLET, J.M. CONIA, *Synthesis*, 1978, 63.
- 5 . T.J. NITZ, L.A. PAQUETTE, *Tetrahedron Lett.* 1984, 25, 3047.
- 6 . All new compounds gave satisfactory analytical and spectral data .  
7a : oil, IR (CCl<sub>4</sub>) 700, 710, 1110, 1715, 3040, 3070, 3540 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 4.25 (s, 2H), 5.16 (s, 1H), 7.30 (s, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 64.8 (t), 77.3 (d), 126.7 (d), 129.0 (d), 137.0 (s), 209 (s) .  
7b : m.p. 39-41°C (petroleum ether-ether); IR (CCl<sub>4</sub>) 1710, 3500, 3620 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.87 (m, 8H), 3.0 (m, 2H), 4.59 (s, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) 24.4 (t), 39.4 (t), 64.9 (t), 86.4 (s), 214.2 (s) .  
7c : m.p. 88-89°C (ether-pentane); IR (CCl<sub>4</sub>) 1710, 3520, 3620 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.61 (m, 10H), 2.73 (m, 2H), 4.51 (s, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 20.5 (t), 24.7 (t), 33.9 (t), 64.2 (t), 77.5 (s), 214.4 (s) .  
7d : m.p. 92-94° (ether); IR (CCl<sub>4</sub>) 1715, 3025, 3380, 3440, 3530, 3640 cm<sup>-1</sup>; <sup>1</sup>H NMR (60MHz, CDCl<sub>3</sub>) δ 1.08-2.41 (m, 7H), 3.06 (m, 2H), 4.21 (m, 1H), 4.40 (br. s, 2H), 5.61 (br. s, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 21.8, 23.7, 24.9, 25.1, 25.5, 27.8, 37.7, 37.9, 66.0, 78.4, 125.2, 125.7, 126.6, 126.7 .  
7e : m.p. 239-241 (methanol); [α]<sub>D</sub> = -40.5 (c = 0.3, pyridine); IR (Nujol) 3480, 3390, 1705, 1090 cm<sup>-1</sup>; <sup>1</sup>H NMR (Py-d5) δ 1.06 (s, 3H), 1.33 (s, 3H), 3.88 (s, 4H), 5.0 (br. s, 2H), 5.36 (m, 1H); <sup>13</sup>C NMR (Py-d5) δ 15.2, 18.1, 24.7, 31.4, 32.3, 34.2, 35.2, 36.2, 37.1, 41.9, 48.1, 50.4, 51.9, 59.7 (C<sub>9</sub>), 64.3 (2C, C<sub>22</sub>, C<sub>23</sub>), 67.8 (C<sub>21</sub>), 88.7 (C<sub>17</sub>), 109.5 (C<sub>3</sub>), 120.9 (C<sub>6</sub>), 140.9 (C<sub>5</sub>), 209.2 (C<sub>20</sub>), 217.5 (C<sub>11</sub>) .  
8 : (unnatural cortisone) : m.p. 179-180 (benzene) ; [α]<sub>D</sub> = + 108 (c = 0.27, abs. ethanol); IR (CHCl<sub>3</sub>) 3420, 1710, 1665, 1615 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.90 (s, 3H), 1.40 (s, 3H), 4.50 (br. s, 2H), 5.70 (br. s, 1H) ; <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.7, 16.9, 23.6, 29.5, 31.6, 32.0, 33.5, 34.2, 36.6, 37.2, 38.0, 47.7, 50.8, 62.0 (C<sub>9</sub>), 67.0 (C<sub>21</sub>), 88.2 (C<sub>17</sub>), 124.3 (C<sub>4</sub>), 169.0 (C<sub>5</sub>), 200.1 (C<sub>3</sub>), 207.8 (C<sub>20</sub>), 215.5 (C<sub>11</sub>) .

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